

## ALIPHATIC COMPOUNDS

**Acetal (acetaldehyde diethylacetal)** [105-57-7] **M 118.2, b 103.7-104°**,  $d_4^{20}$  **0.831**,  $n_D^{25}$  **1.38054**,  $n_D^{25}$  **1.3682**. Dry acetal over Na to remove alcohols and H<sub>2</sub>O, and to polymerise aldehydes, then fractionally distil. Or, treat it with alkaline H<sub>2</sub>O<sub>2</sub> at 40-45° to remove aldehydes, then saturate with NaCl, separate, dry with K<sub>2</sub>CO<sub>3</sub> and distil it from Na [Vogel *J Chem Soc* 616 1948]. [Beilstein 1 IV 3103.]

**Acetaldehyde** [75-07-0] **M 44.1, b 20.2°**,  $d_4^{20}$  **0.788**,  $n_D^{25}$  **1.33113**, **pK<sup>25</sup> 13.57 (hydrate)**. Acetaldehyde is usually purified by fractional distillation in a glass helices-packed column under dry N<sub>2</sub>, discarding the first portion of distillate. Or, it is shaken for 30minutes with NaHCO<sub>3</sub>, dried with CaSO<sub>4</sub> and fractionally distilled at 760mm through a 70cm Vigreux column (p 11). The middle fraction is collected and further purified by standing for 2hours at 0° with a small amount of hydroquinone (free radical inhibitor), followed by distillation [Longfield & Walters *J Am Chem Soc* 77 810 1955]. [Beilstein 1 IV 3094.]

**Acetaldehyde dimethyl acetal (1,1-dimethoxyethane)** [534-15-6] **M 90.1, b 63-65°**,  $d_4^{20}$  **0.852**,  $n_D^{25}$  **1.36678**. Distil the dimethyl acetal through a fractionating column and fraction boiling at 63.8°/751mm is collected. It forms an azeotrope with MeOH. Alternatively purify it as for *acetal* above. It has been purified by GLC. [Beilstein 1 IV 3103.]

**Acetamide** [60-35-5] **M 59.1, m 81°**, **pK<sub>1</sub><sup>25</sup> -1.4, pK<sub>2</sub><sup>25</sup> +0.37**. Acetamide is crystallised by dissolving in hot MeOH (0.8mL/g), diluting with Et<sub>2</sub>O and allowing to stand [Wagner *J Chem Edu* 7 1135 1930]. Alternate crystallisation solvents are acetone, \*benzene, chloroform, dioxane, methyl acetate or \*benzene/ethyl acetate mixtures (3:1 and 1:1). It has also been recrystallised from hot water after treating with HCl-washed activated charcoal (which had been repeatedly washed with water until free from chloride ions), then crystallised again from hot 50% aqueous EtOH and finally twice from hot 95% EtOH [Christoffers & Kegeles *J Am Chem Soc* 85 2562 1963]. Finally it is dried in a vacuum desiccator over P<sub>2</sub>O<sub>5</sub>. Acetamide is also purified by distillation (**b** 221-223°) or by sublimation *in vacuo*. It has also been purified by two recrystallisations from cyclohexane containing 5% (v/v) of \*benzene. Needle-like crystals separate and are filtered, washed with a small volume of distilled H<sub>2</sub>O and dried with a flow of dry N<sub>2</sub>. [Slebocka-Tilk et al. *J Am Chem Soc* 109 4620 1987, Beilstein 2 H 175, 2 I 80, 2 II 177, 2 III 384, 2 IV 399.]

**Acetamidine hydrochloride** [124-42-5] **M 94.5, m 164-166°, 165-170°(dec), 174°**, **pK<sup>25</sup> 12.40**. The hydrochloride can be recrystallised from small volumes of EtOH. Alternatively it is dissolved in EtOH, filtered, Et<sub>2</sub>O is added; filter the crystalline salt off under N<sub>2</sub> and dry it in a vacuum desiccator over H<sub>2</sub>SO<sub>4</sub>. The salt is deliquescent and should be stored in a tightly stoppered container. Its solubility in H<sub>2</sub>O is 10% at room temperature and it is soluble in Me<sub>2</sub>CO. The *free base* reacts strongly alkaline in H<sub>2</sub>O. It has  $\lambda_{max}$  224nm ( $\epsilon$  4000) in H<sub>2</sub>O. The *picrate* has **m** 252° (sintering at ~245°). [Dox *Org Synth Coll Vol I* 5 1941, Davies & Parsons *Chem Ind (London)* 628 1958, Barnes et al. *J Am Chem Soc* 62 1286 1940 give **m** 177-178°, Beilstein 2 H 185, 2 I 85, 2 II 183, 2 III 416, 2 IV 428.]

**N-(2-Acetamido)-2-aminoethanesulfonic acid (ACES)** [7365-82-4] **M 182.2, m > 220°(dec)**, **pK<sub>Est</sub> ~1.5, pK<sub>2</sub> 6.9**. Recrystallise ACES from hot aqueous EtOH. [Perrin & Dempsey *Buffers for pH and Metal Ion Control* Chapman & Hall, London 1974, Beilstein 4 III 1707.]

**N-(2-Acetamido)iminodiacetic acid (ADA)** [26239-55-4] **M 190.2, m 219°(dec)**, **pK<sub>1</sub> ~2.3, pK<sub>2</sub> 6.6**. Dissolve ADA in water, add one equivalent of NaOH solution (to final pH of 8-9), then acidify with HCl to precipitate the free acid. This is filtered off, washed with water and dried *in vacuo*. [Beilstein 4 IV 2441.]

**Acetamidomethanol** [625-51-4] **M 89.1, m 47-50°, 54-56°, 55°**. Recrystallise it from freshly distilled Me<sub>2</sub>CO, wash the crystals with dry Et<sub>2</sub>O and dry them in a vacuum desiccator over P<sub>2</sub>O<sub>5</sub>. **R<sub>F</sub> 0.4** on paper chromatography with CHCl<sub>3</sub>/EtOH (2:8) as solvent and developed with ammoniacal AgNO<sub>3</sub>. It also crystallises in needles from EtOAc containing a few drops of Me<sub>2</sub>CO. It is *hygroscopic* and should be stored under dry conditions. [Bachmann et al. *J Am Chem Soc* 73 2775 1951, Walter et al. *Chem Ber* 99 3204 1966, Einhorn & Ladisch *Justus Liebig's Ann Chem* 343 265 1905, Beilstein 2 IV 405.]